
SCIENCE

1113/02

Paper 2

October 2018

MARK SCHEME

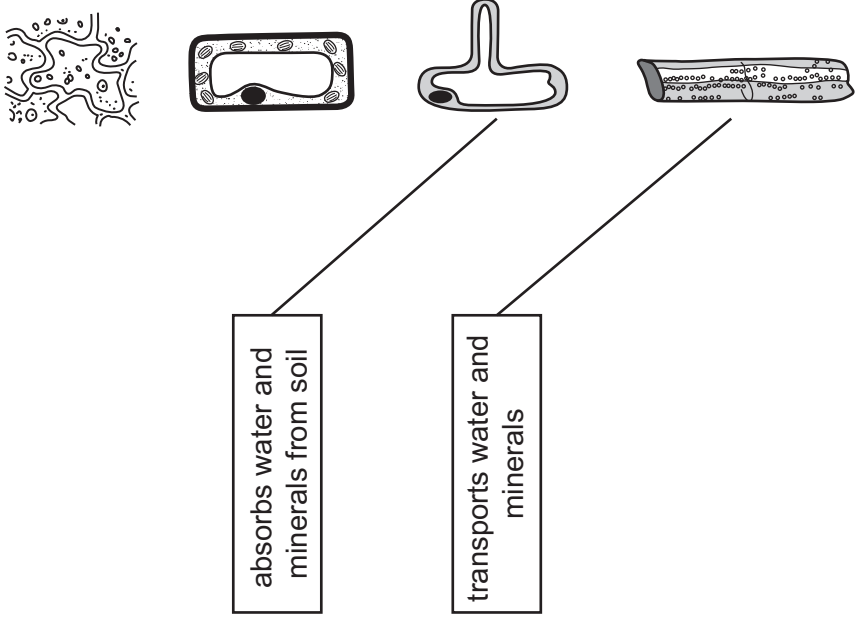
Maximum Mark: 50

Published

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Markers were instructed to award marks. It does not indicate the details of the discussions that took place at an Markers' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the End of Series Report. Cambridge will not enter into discussions about these mark schemes.

This document consists of **11** printed pages.

Question	Answer	Marks	Further Information
1	<p>function</p> <p>plant cell</p>  <p>The diagram shows a plant cell with several internal structures. Two boxes with lines pointing to specific parts contain the following text:</p> <ul style="list-style-type: none"> absorbs water and minerals from soil transports water and minerals 	2	<p>each correct line = 1 mark</p> <p>two lines from a function = 0 marks</p>

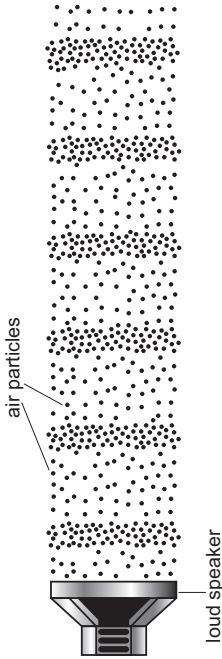
Question	Answer	Marks	Further Information								
2(a)	<table border="1"> <tr> <td>non-metal</td> <td>type of compound</td> </tr> <tr> <td>chlorine</td> <td>chloride</td> </tr> <tr> <td>oxygen</td> <td>oxide</td> </tr> <tr> <td>sulfur</td> <td>sulfide</td> </tr> </table>	non-metal	type of compound	chlorine	chloride	oxygen	oxide	sulfur	sulfide	2	
non-metal	type of compound										
chlorine	chloride										
oxygen	oxide										
sulfur	sulfide										
2(b)	water	1									
Question	Answer	Marks	Further Information								
3	<p>A₂ is 0.4 A.</p> <p>A₃ is 0.4 A.</p> <p>A₄ is 0.8 A.</p>	2	<p>three correct = 2 marks</p> <p>two correct = 1 mark</p> <p>one correct = 0 marks</p>								

Question	Answer	Marks	Further Information
4	<p>frog = amphibia(n)</p> <p>valid reason e.g. larval stage / smooth skin / moist skin</p>	2	<p>Note second marking point is dependent on correct identification of amphibian(n)</p> <p>Accept reference to young being tadpoles sufficient for larval stage</p> <p>Accept permeable skin / idea of absorbing oxygen through the skin / lays eggs in water / lays unshelled eggs</p>

Question	Answer	Marks	Further Information
5(a)	<p>any one from</p> <p>different size stars</p> <p>different brightness of stars</p> <p>stars not in the same position</p>	1	<p>Accept the biggest star has moved</p> <p>Accept stars have moved / position of stars / constellations have moved / arrangement of stars is different</p>
5(b)	same or similar to the one in January	1	

Question	Answer	Marks	Further Information
6(a)(i)	photosynthesis	1	
6(a)(ii)	oxygen	1	
6(b)	gas A = grown / increased in size gas B = small(er) / dying gas C = (idea of) no change / not grown	3	Accept grew very well / tall(er) Accept wilting Accept same size

Question	Answer	Marks	Further Information
7(a)	40 (cm ³)	1	
7(b)	any volume greater than 22 (cm ³)	1	
7(c)	more particles more collisions	2	Accept less space between particles / more particles in one unit volume / more crowded particles / more particles in the same space / particles closer together / more particles (to) collide / particles more concentrated / more frequent collisions / collisions more often / more chance of collisions = 2 marks

Question	Answer	Marks	Further Information
8	 <p>air particles</p> <p>loud speaker</p> <p>letter C touching area where dots are close together</p> <p>letter R touching area where dots are far apart</p>	2	

Question	Answer	Marks	Further Information
9(a)	(aerobic) respiration	1	Accept anaerobic respiration Do not accept breathing
9(b)	oxygen	1	
9(c)	carbon dioxide	1	
9(d)	any two from air is less dense / few(er) particles / particles further apart so less oxygen (taken in with each breath idea) lower concentration gradient (for gaseous exchange) so gaseous exchange is less efficient / diffusion (of oxygen) is slower	2	

Question	Answer	Marks	Further Information		
12(a)	<table border="1" style="width: 100%; border-collapse: collapse;"> <tr> <td style="width: 50%; padding: 5px;"> <p style="text-align: center;">compound</p> <p style="text-align: center;">magnesium carbonate</p> <p style="text-align: center;">water</p> <p style="text-align: center;">.....</p> <p style="text-align: center;">.....</p> </td> <td style="width: 50%; padding: 5px;"> <p style="text-align: center;">element</p> <p style="text-align: center;">(carbon)</p> <p style="text-align: center;">copper</p> <p style="text-align: center;">sulfur</p> <p style="text-align: center;">.....</p> </td> </tr> </table>	<p style="text-align: center;">compound</p> <p style="text-align: center;">magnesium carbonate</p> <p style="text-align: center;">water</p> <p style="text-align: center;">.....</p> <p style="text-align: center;">.....</p>	<p style="text-align: center;">element</p> <p style="text-align: center;">(carbon)</p> <p style="text-align: center;">copper</p> <p style="text-align: center;">sulfur</p> <p style="text-align: center;">.....</p>	2	<p>all correct = 2 marks</p> <p>two or three correct = 1 mark</p> <p>one correct = 0</p> <p>note if substance is written in both columns then that substance is incorrect</p>
<p style="text-align: center;">compound</p> <p style="text-align: center;">magnesium carbonate</p> <p style="text-align: center;">water</p> <p style="text-align: center;">.....</p> <p style="text-align: center;">.....</p>	<p style="text-align: center;">element</p> <p style="text-align: center;">(carbon)</p> <p style="text-align: center;">copper</p> <p style="text-align: center;">sulfur</p> <p style="text-align: center;">.....</p>				
12(b)	<p>Mixtures are pure substances. <input style="width: 50px;" type="checkbox"/></p> <p>Mixtures can be separated by physical means. <input checked="" style="width: 50px;" type="checkbox"/></p> <p>Mixtures only contain elements. <input style="width: 50px;" type="checkbox"/></p> <p>Elements in mixtures are joined by chemical bonds. <input style="width: 50px;" type="checkbox"/></p>	1	<p>more than one tick ✓ = 0 marks</p>		
12(c)	nitric (acid)	1			

Question	Answer	Marks	Further Information
13	<p>All the water molecules in the liquid are moving.</p> <p>Some of the molecules have more energy.</p> <p>These molecules move fast / quick / rapid enough to escape the surface of the liquid.</p> <p>This is called evaporation.</p> <p>This makes the liquid become cooler.</p>	4	<p>each correct answer = 1 mark</p> <p>Accept heat energy / speed</p> <p>Ignore freely</p> <p>Accept vigorous</p> <p>Accept denser / decrease in volume / gas / (water) vapour</p>

Question	Answer	Marks	Further Information								
14(a)	<table border="1"> <thead> <tr> <th>concentration of carbon dioxide</th> <th>concentration of oxygen</th> </tr> </thead> <tbody> <tr> <td>high</td> <td>high</td> </tr> <tr> <td>low</td> <td>low</td> </tr> <tr> <td>✓</td> <td>✓</td> </tr> </tbody> </table>	concentration of carbon dioxide	concentration of oxygen	high	high	low	low	✓	✓	1	<p>both ticks required</p> <p>any incorrect tick = 0 marks</p>
concentration of carbon dioxide	concentration of oxygen										
high	high										
low	low										
✓	✓										
14(b)(i)	artery	1									
14(b)(ii)	(blood is at high) pressure / so it does not rupture	1	Accept so blood vessel will not burst / resists the pressure of the blood								

Question	Answer	Marks	Further Information
15(a)	A B C D E	1	more than one answer = 0 marks
15(b)	increasing / going up	1	Accept accelerating / (goes) faster / slow then goes fast
Question	Answer	Marks	Further Information
16	force x distance is a moment balance when moments are equal / balance when clockwise moment equals anti-clockwise moment but for 2 marks $F_1 \times d_1 = F_2 \times d_2$	2	Accept $F_1 \times d_1$ or $F_2 \times d_2$ answer must refer to moments rather than force or pressure