

Cambridge Lower Secondary Checkpoint

SCIENCE 1113/01

Paper 1 April 2020

MARK SCHEME

Maximum Mark: 50

Published

This mark scheme is published as an aid to teachers and learners, to indicate the requirements of the examination. However, we have not been able to adjust it to reflect the full range of answers that would have been seen as a part of the normal moderation and marking process, and it does not necessarily contain all the possible alternatives that might have arisen.

Cambridge will not enter into discussions about the mark scheme.

General guidelines on marking

Many descriptive answers can be expressed in a variety of ways. Professional judgement can be used in these cases, providing it matches the marking points and further information in the mark scheme.

Answers may have words spelt incorrectly. Credit is normally given for phonetically correct answers, unless the word has a scientifically different meaning. For example, where the answer should be antennae, credit will be given for antena but not for anthen (too close to anther).

Only the science is being assessed so answers do not need to be grammatically correct. Significant figures will be indicated in the question or in the mark scheme.

Unless specified all marking points are independent.

Annotations and abbreviations

/OR alternate responses for the same marking point

() brackets the words or units in brackets do not need to be stated, for example, (recycles or

releases or provides) minerals = minerals scores the mark

Underline exact word is required

Accept an acceptable response

Do not accept indicates an incorrect response that would contradict another otherwise correct

alternative

Ignore indicates an irrelevant answer that is not creditworthy. Full marks can still be

achieved even with answers that are ignored.

Note provides extra information when necessary

ecf error carried forward: marks are awarded if an incorrect response has been carried

forward from earlier working, provided the subsequent working is correct

ora or reverse argument; for example, as mass increases, volume increases could be

written as mass decreases, volume decreases

Question			Answe	r	Marks	Further Information
1	part of a cell cell membrane cell wall chloroplast	only in animal cells	only in plant cells	in both animal and plant cells	2	all three correct = 2 marks two correct = 1 mark one correct = 0 marks if more than one tick in a row, 0 marks for that row

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Question	Answer	Marks	Further Information
2(a)	sonorous bridge across a river strong electrical wires bell hard	2	all four correct = 2 marks two or three correct = 1 mark one correct = 0 marks
2(b)	any two from plastic is an insulator/does not conduct electricity (idea of) makes the wires safe/prevents electrocution flexible can be coloured	2	each correct answer = 1 mark Accept other valid answers
2(c)	All metals have low melting points. Some metals are gases at room temperature. All metals conduct heat.	1	more than one box ticked = 0 marks

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Question	Answer	Marks	Further Information
3(a)	Sun	1	
3(b)	any two from	2	each correct answer = 1 mark
	planets move around the Sun/planets do not move around the Earth/only the Moon moves around the Earth		
	the Earth moves/the Sun does not move		
	(idea that) not all the planets are in the model		Accept named planets missing e.g. Uranus is not in the model
	the order of the planets is not correct		Uranus is not in the model
	orbit size is not correct/not to scale		
	objects in model are not to scale		
3(c)	any one from	1	
	the Moon moves around the Earth		
	planets move in orbits		
	a correct statement about the order of the planets e.g.		
	a correct statement about the relative size of the planets e.g. Jupiter is the largest planet		

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Question	Answer	Marks	Further Information
4(a)	A	2	all three correct = 2 marks
			one or two correct = 1 mark
	В		
	C		
4(b)	A = artery	2	Accept aorta/named artery
	C = vein		Accept vena cava/named vein

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Question	Answer	Marks	Further Information
5(a)	chlorine and bromine	1	both correct = 1 mark
			Accept liquid and gas
5(b)	Cl	1	Accept Cl ₂
5(c)	Forces between the particles push them apart. The particles are free to move. The particles can easily be squashed into a small space. The particles increase in size to fill the space.	1	more than one box ticked = 0 marks
5(d)	any two from particles gain energy/(some) particles have more kinetic energy/(some) particles move faster particles have overcome attraction between them/idea that forces between particles have been broken particles have spread further apart (in the gas phase)	2	Accept molecules instead of particles Do not accept boils Accept bromine changes to a gas/ bromine changes to vapour Accept particles diffuse into the air/ particles escape from the liquid

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Question	Answer	Marks	Further Information
6(a)	vibrating/moving (air) particles	2	Accept rarefaction and compression as an alternative to vibrating
6(b)(i)	any wave with greater amplitude	1	At least one wave greater than 6 small squares. Above or below the midline Ignore frequency
6(b)(ii)	any wave with greater frequency	1	Diagram must show at least two complete waves Accept incomplete waves providing at least two complete waves are drawn Ignore amplitude

Question	Answer			Marks	Further Information
7(a)	(seed germination) → pollination → fertilisation → seed formation			1	all three in correct order for one mark
7(b)				2	each method and reason correctly
		method of seed dispersal	reason		linked = 1 mark
	Α	by animals	have hooks/attach to hair or fur		
	В	by wind	light (weight)/feathery/ parachute shaped		

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Question	Answer	Marks	Further Information
8(a)(i)	3	1	
8(a)(ii)	4	1	
8(b)	2.8.1 or	2	If drawn ignore distribution of electrons within each shell so long as numbers are correct Accept dots or circles instead of crosses total of 11 electrons drawn outside of the nucleus = 1 mark But drawing showing 2.8.1 = 2 marks Accept correct use of 2.8.1 notation = 2 marks Ignore any particles shown in the nucleus

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Question	Answer	Marks	Further Information
9(a)(i)	both picked up 24/both picked up the maximum number of paper clips	1	
9(a)(ii)	using more paper clips	1	Accept repeat using bigger paper clips /replace with iron filings and measure the mass of iron filings
9(b)(i)	fewer paper clips picked up	1	
9(b)(ii)	their hands are different/Pierre's hand is bigger/Mike's hand is smaller	1	

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Question	Answer	Marks	Further Information
10(a)(i)	answer in the range pH 2 – pH 2.5	1	
10(a)(ii)	any two from	2	each correct answer = 1 mark
	test at more pH values		
	test between pH 1 and pH 3/test at pH 2		
	repeat the experiment/take averages		
10(b)(i)	any one from	1	
	liquid is corrosive/burn tissue/liquid would harm skin		
10(b)(ii)	any one from	1	Accept avoid contact with skin/avoid
	wear goggles/wear gloves/apron/lab coat		contact with eyes/do not swallow/do not ingest it/mop up spillages
	do not taste/smell the liquid		
	use only small quantities		
10(c)	any one from	1	
	(keep same) temperature		
	(use same) amount or proportion or concentration of enzyme		
	(use same) amount or proportion or concentration of egg white		
	mix liquids evenly		

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Question	Answer	Marks	Further Information
11(a)	colour change/fizzing/bubbles/new substance forms/some form of coating	1	Accept temperature change
11(b)(i)	magnesium nitrate x x x	1	Ignore X in the grey box
11(b)(ii)	copper	1	

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Question	Answer	Marks	Further Information
12(a)	see diagram below reflection	1	reflected ray correct by eye = 1 mark Do not accept arrow in the wrong direction note – arrow does not need to be shown
12(b)	air water	1	refracted ray correct by eye = 1 mark Ignore arrows
12(c)	see diagram below screen light source opaque object	2	two rays from light source continued to edge of object and onto screen = 1 mark middle ray goes to object = 1 mark Ignore arrows

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Question	Answer	Marks	Further Information
13	(glucose) + oxygen → water + carbon dioxide	2	oxygen as a reactant = 1 mark
			water and carbon dioxide (either order) as products = 1 mark
			Accept formula but names take precedence
			 oxygen, O₂ water, H₂O carbon dioxide, CO₂

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